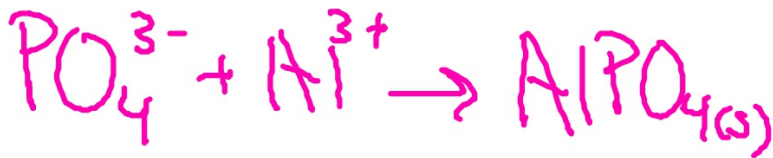
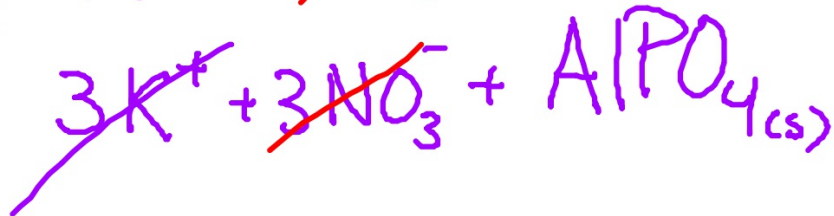
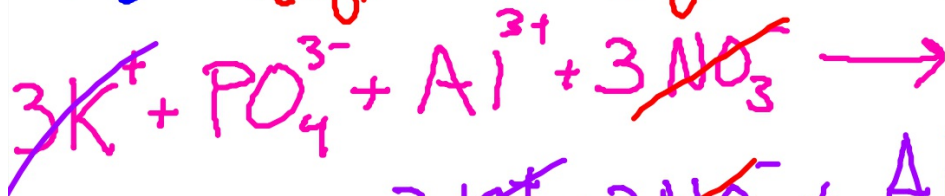


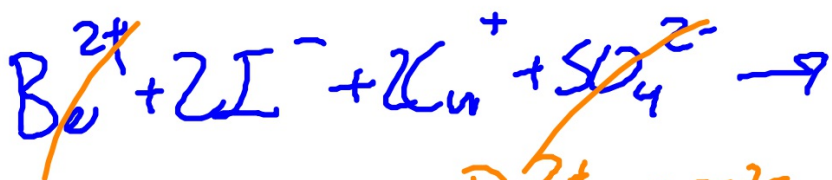
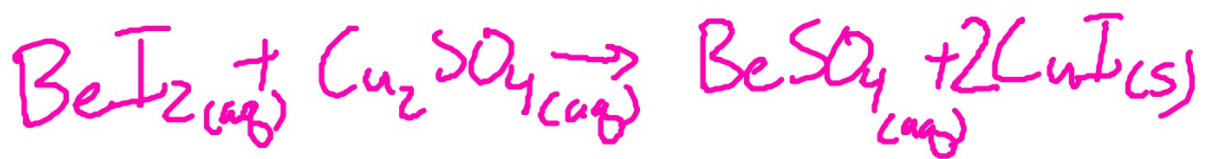
Net ionic equations

1.) Break up everything that
is not a (s), (l), or (g)
Write as an ion

2.) Cancel out anything that
is the same on both sides
of the yield sign

3.) Re-write





Calcium Hydroxide + Iron(III) Chloride \rightarrow Iron(III) Hydroxide(s) + Calcium Chloride

