BO₂H₃

HOF

3. How could this Lewis structure be better?

• SO_3^{2-}

C₂H₄

• Draw 2 resonance structures for the nitrite ion.

• NO_2^{1-}

7. Draw two Lewis structures of the following:

SO₃

(One structure should have sulfur with an octet and the other should have it with an expanded octet)

8. Draw the Lewis structure of the following:

CH₂Br₂

(carbon is the central atom)

9. Draw the Lewis structure of the following:

BCl₃

10. Draw the Lewis structure of the following and name the ion:

 $H_2PO_4^{1-}$

(H's are attached to O's)

11. Draw the Lewis structure of the following and name the ion:

HCO₃¹⁻

(H is attached to O)

CH₄O